

# Multiple Choice Thermochemistry Questions

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- The value of  $\Delta H^\circ$  for the reaction below is 128 kJ  $\text{CH}_3\text{OH}(\text{l}) \rightarrow \text{CO}(\text{g}) + 2\text{H}_2(\text{g})$  How many kJ of heat are consumed when 15.5 g of  $\text{CH}_3\text{OH}(\text{l})$  decomposes as

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